

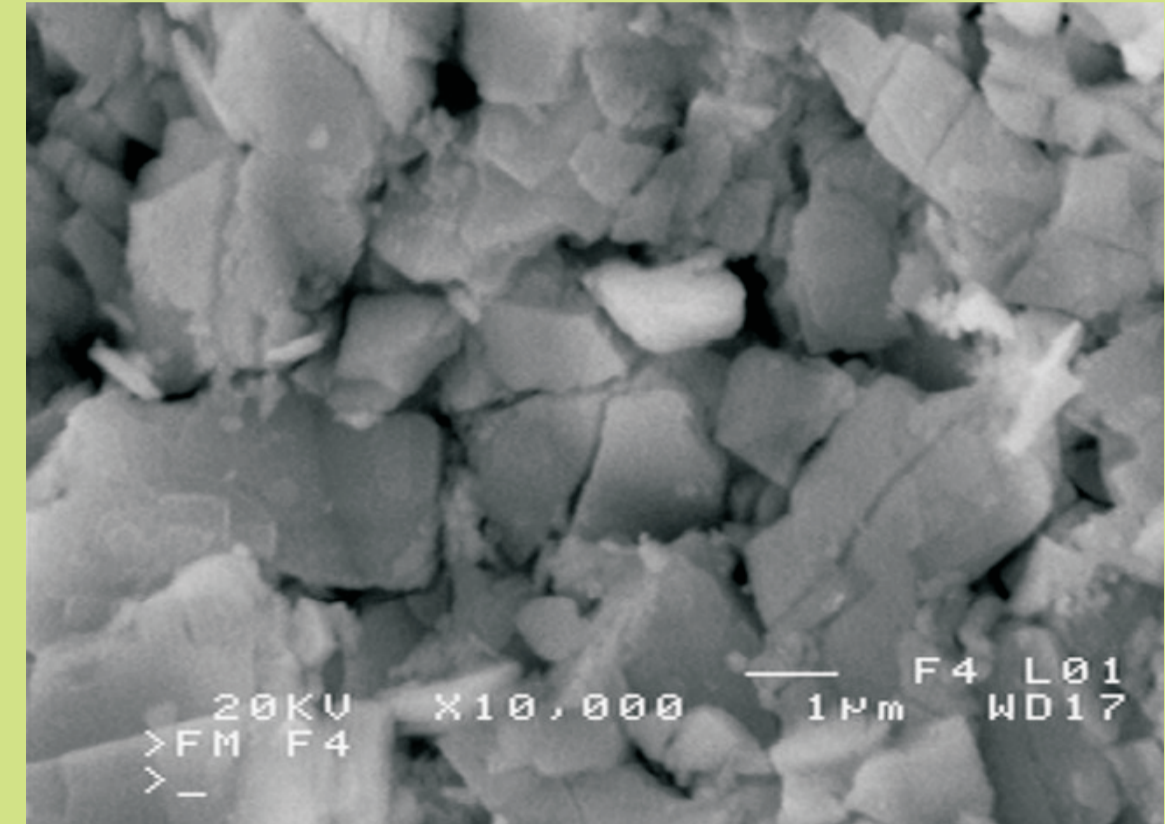
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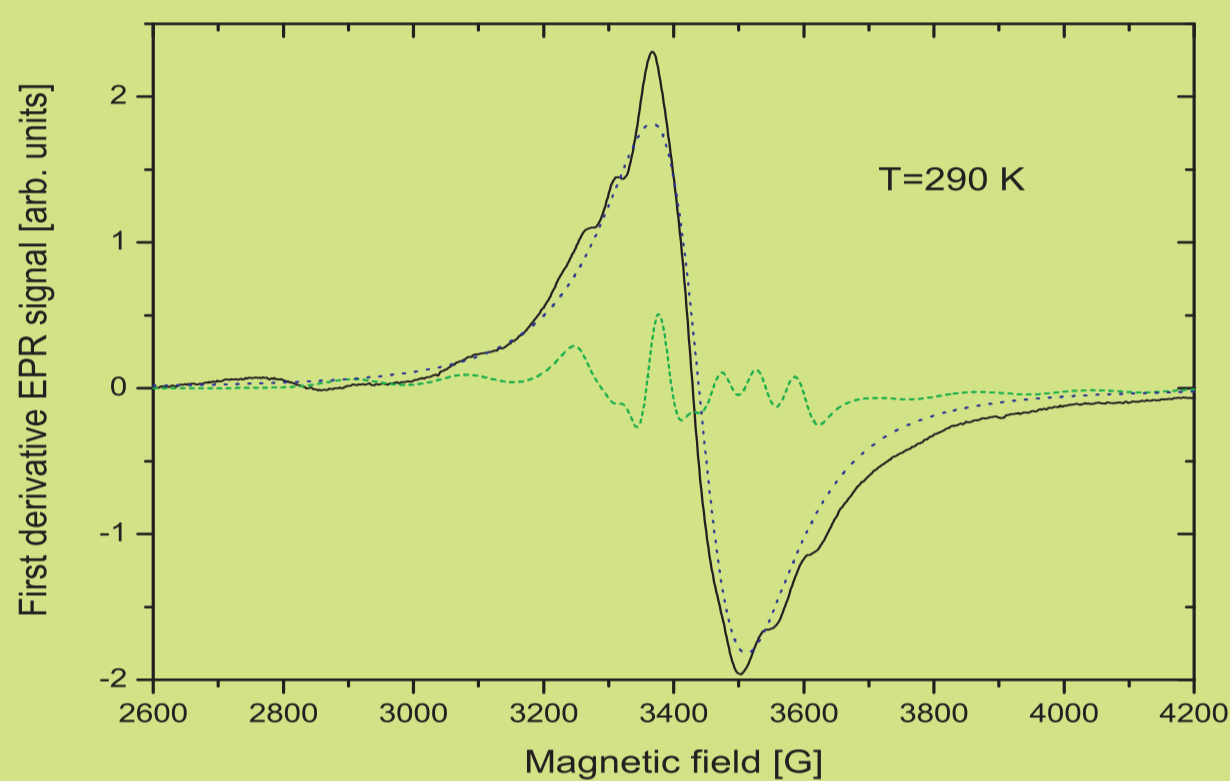
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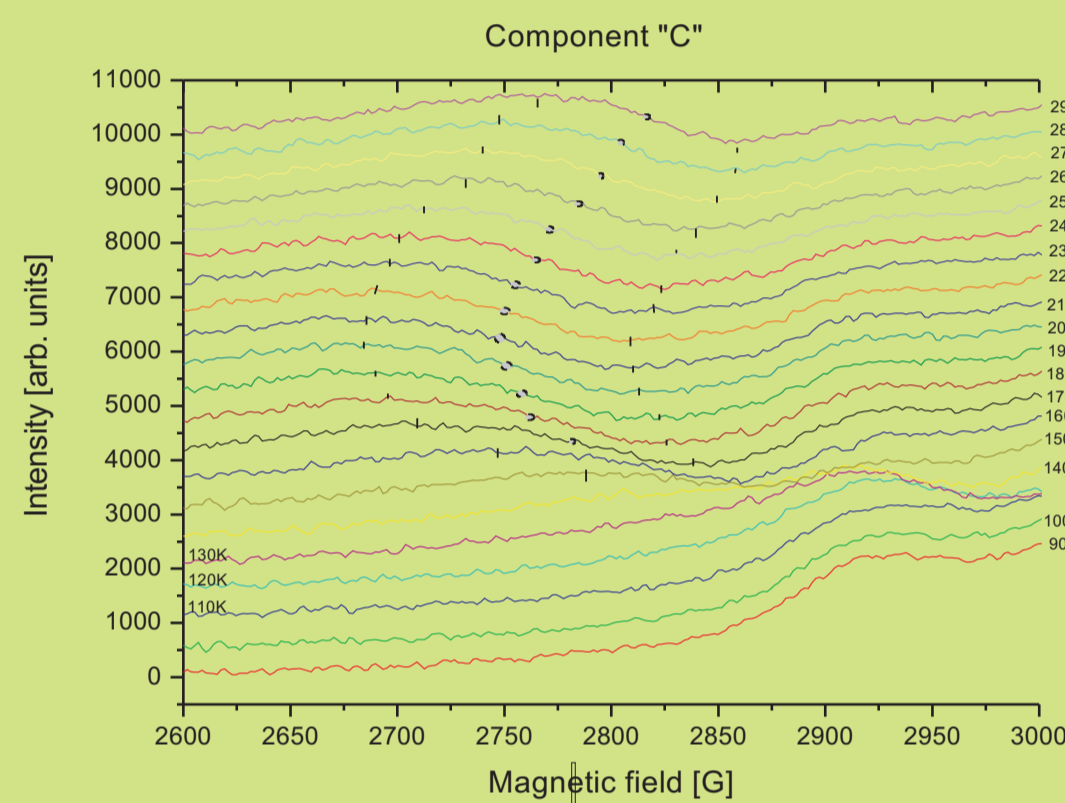
In the last years, the research on multicomponent oxides systems has been intensified in order to obtain new materials of interesting electric, magnetic or catalytic properties. The basic studies on the ternary $\text{V}_2\text{O}_5\text{-Nb}_2\text{O}_5\text{-Sb}_2\text{O}_4$ system have shown that the oxides formed in the solid state in air contain an unknown compound with the formula $\text{Nb}_2\text{VSbO}_{10}$. This compound has been obtained by heating of the mixture $\text{V}_2\text{O}_5/\text{Sb}_2\text{O}_4/\text{Nb}_2\text{O}_5$ (in molar ratio 1:1:2) in air at temperatures up to 750°C according to the equation: $\text{V}_2\text{O}_5 + 2\text{Nb}_2\text{O}_5 + \text{Sb}_2\text{O}_4 + 1/2\text{O}_2(\text{g}) = 2\text{Nb}_2\text{SbVO}_{10(\text{s})}$. The compound $\text{Nb}_2\text{VSbO}_{10}$ can be also obtained by heating an equimolar mixture of SbVO_5 and $\text{T-Nb}_2\text{O}_5$ according to the equation: $\text{Nb}_2\text{O}_5(\text{s}) + \text{SbVO}_5(\text{s}) = \text{Nb}_2\text{SbVO}_{10(\text{s})}$. The new compound $\text{Nb}_2\text{VSbO}_{10}$ is stable in air up to 880°C and next melts incongruently with deposition of solid $\text{Nb}_9\text{VO}_{25}$. It crystallizes in the orthorhombic system with the following unit cell parameters: $a = 0.328143$ nm, $b = 0.458946$ nm, $c = 1.22476$ nm, ($Z = 1$).



SEM picture of $\text{Nb}_2\text{VSbO}_{10}$ compound.



The observed spectrum of $\text{Nb}_2\text{VSbO}_{10}$ (continuous line) at $T=290$ K and its two main components: isolated V^{4+} ions (dashed line - component A) and a broad line (dotted line - component B).



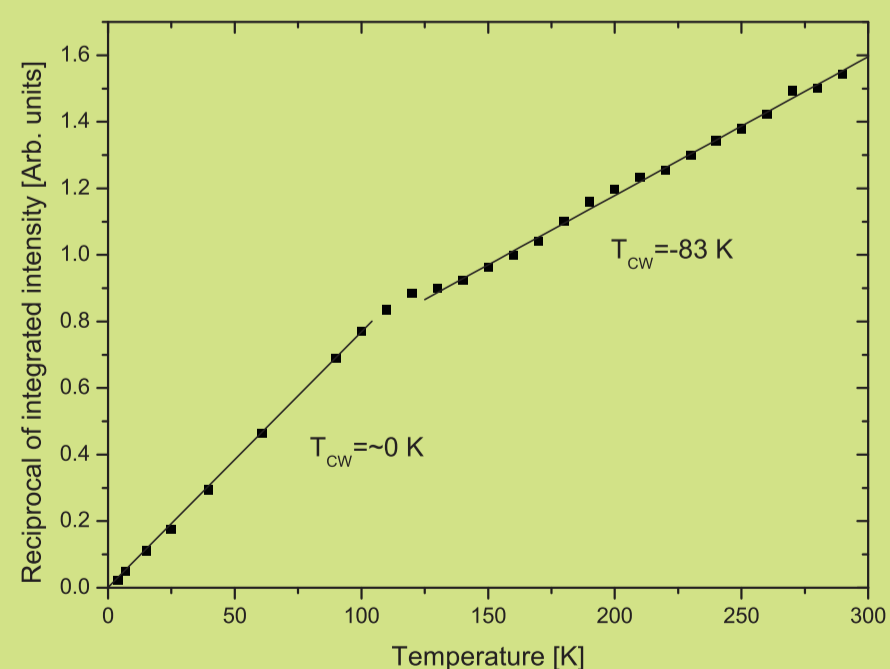
The component C in the EPR spectrum of $\text{Nb}_2\text{VSbO}_{10}$ at different temperatures.

Electron paramagnetic resonance (EPR) study of $\text{Nb}_2\text{VSbO}_{10}$ has been carried out on a conventional X-band continuous wave Bruker E 500 equipped with TE_{102} cavity with 100 kHz modulation. The temperature variation studies in the 7-300 K range were performed using an Oxford ESP300 continuous flow helium cryostat.

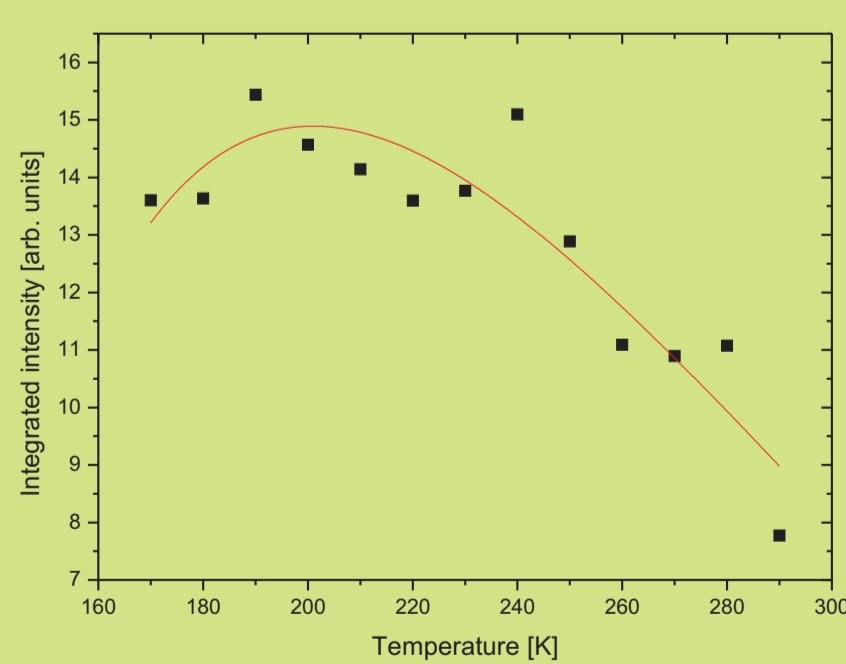
To estimate the number of spins in $\text{Nb}_2\text{VSbO}_{10}$ sample participating in the resonance a comparative study with standard $\text{VO}_2 \cdot 5\text{H}_2\text{O}$ sample with a known number of spins has been performed. Assuming that vanadium ions in the $\text{Nb}_2\text{VSbO}_{10}$ compound could be only in 4+ (EPR active) or 5+ (EPR inactive) valence states it was calculated that only 5.2% of vanadium ion are magnetic and thus in 4+ valence state. This concentration of magnetic ions is even smaller than in previously studied SbVO_5 compound. The knowledge of concentration of magnetic ions might be important in estimation of the catalytic activity of the compound as often the minority magnetic ions play a dominant role in such activity.

The EPR spectrum of $\text{Nb}_2\text{VSbO}_{10}$ in the high temperature range could be decomposed on three components: a broad signal in the range of $g \approx 2$ without hyperfine structure (hfs), a partially resolved hfs lines typical of isolated vanadium ions in an axial symmetry and a weak line near 2800 G. The broad line without hfs could be attributed to a mobile electron hopping along the $\text{V}^{4+}\text{-O-V}^{5+}$ bond. The hfs could be substantially suppressed or even disappear due to various interactions of electronic spins with their surroundings. One such interaction occurs via the super-exchange through oxygen bridge of an electron between aliovalent vanadium centers. The second component displays usually two sets of eight lines, partially overlapping, attributed to the interaction of electron spin ($S=1/2$) with ^{51}V nucleus ($I=7/2$, abundance 99.75%). This spectrum could be satisfactorily described by an axial spin Hamiltonian of the form $H = [g B_z S_z + g (B_x S_x + B_y S_y)] + A S_z I_z + A (S_x I_x + S_y I_y)$, where the symbols have their usual meaning. Fitting the registered spectrum by using the program SimFonia allowed to obtain the following values of the spin Hamiltonian: $g = 1.907$, $g = 1.947$, $A = 164.6$ G, $A = 43.5$ G. These values didn't show any significant temperature dependence. The third component displays a very anisotropic behaviour. It shifts significantly with temperature, reaching the lowest resonance field of 2800 G at 200 K.

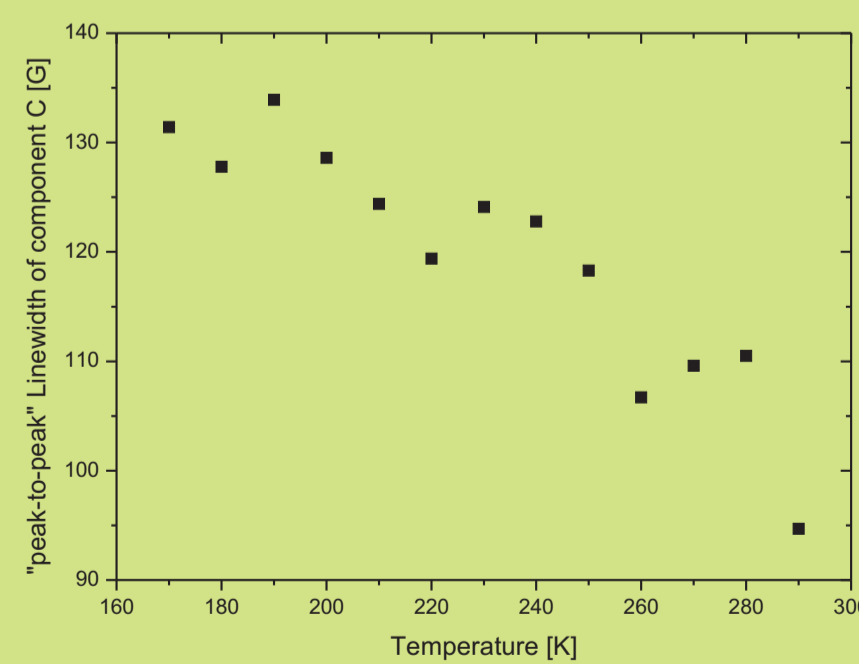
To reveal magnetic interactions in $\text{Nb}_2\text{VSbO}_{10}$ the temperature dependence of EPR integrated intensity has been studied. In the temperature dependence of reciprocal of integrated intensity two temperature regimes are observed: in high temperature range ($T > 120$ K) this dependence is described by a Curie-Weiss law, $I = C/(T - T_0)$, with $T_0 = -82.8$ K, while in low temperature EPR range (below 120 K) the Curie law is registered. Negative value of T_0 indicates on an effective antiferromagnetic interaction in the magnetic system of $\text{Nb}_2\text{VSbO}_{10}$ at high temperature.



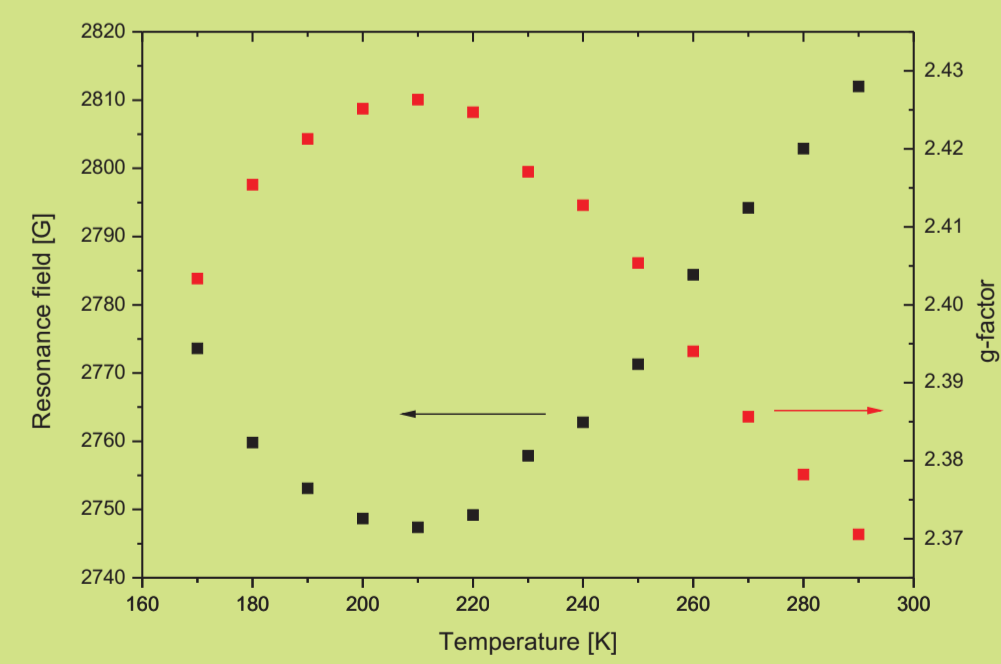
Temperature dependence of the reciprocal integrated intensity of $\text{Nb}_2\text{VSbO}_{10}$. The straight lines are the least squares best fits for the low and high temperature ranges.



Temperature dependence of the integrated intensity of the component C.



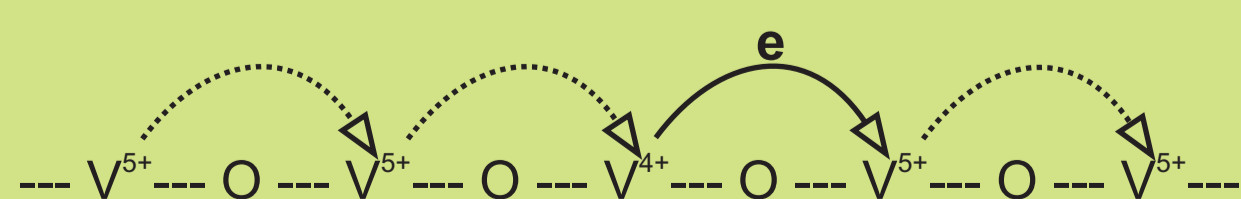
Temperature dependence of peak-to-peak linewidth of the component C.



Temperature dependence of the resonance field (black) and g-factor (red) for the component C.



Model of the component A - isolated V^{4+} ions with hfs.



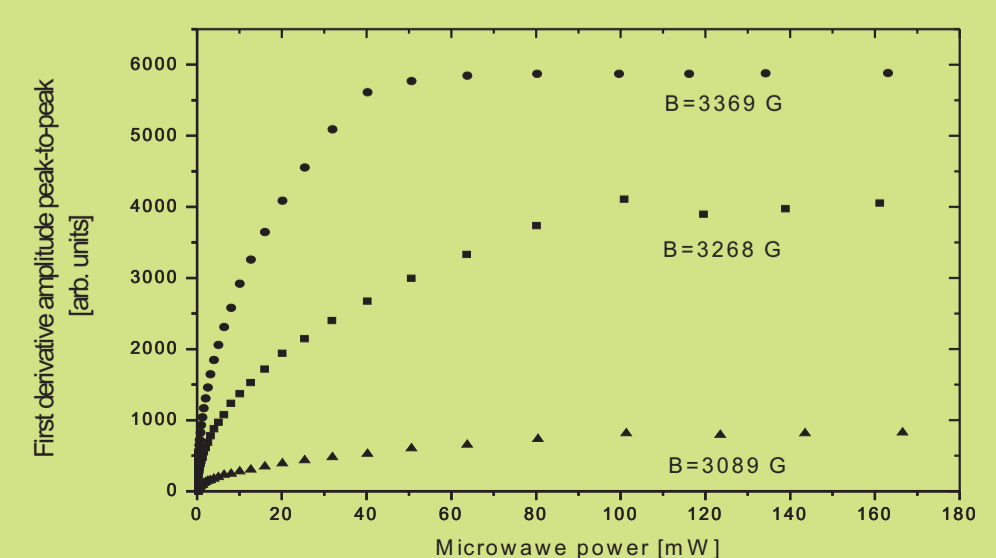
Model of the component B - electron jumping along the vanadium chains.



Model of the component C - antiferromagnetic V^{4+} - V^{4+} dimers

Parameter	Unit	SbVO_5	$\text{Nb}_2\text{VSbO}_{10}$
$g_{ }$	-	1.9311	1.902
g_{\perp}	-	1.9425	1.93725
g_0	-	1.9387	1.9255
$\Delta g_{ } / \Delta g_{\perp}$	-	1.19	1.542
$ A_{ } $	10^{-4} cm^{-1}	181	147.1
$ A_{\perp} $	10^{-4} cm^{-1}	54	38.61
P	10^{-4} cm^{-1}	140	137.1
k	-	0.6244	0.3866
$P \cdot k$	10^{-4} cm^{-1}	87.4	53.01

Comparison of hfs parameters of two similar compounds: SbVO_5 and $\text{Nb}_2\text{VSbO}_{10}$.



EPR signal amplitude as a function of the microwave power for three lines in EPR spectrum of $\text{Nb}_2\text{VSbO}_{10}$.